# Lowering Extracellular Sodium or pH Raises Intracellular Calcium in Gastric Cells

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Summary. The dependence of cytoplasmic free [Ca] (Ca<sub>i</sub>) on [Na] and pH was assessed in individual parietal cells of intact rabbit gastric glands by microfluorimetry of fura-2. Lowering extracellular [Na] (Na<sub>o</sub>) to 20 mm or below caused a biphasic Ca<sub>i</sub> increase which consisted of both release of intracellular Ca stores and Ca entry across the plasma membrane. The Ca increase was not blocked by antagonists of Ca-mobilizing receptors (atropine or cimetidine) and was independent of the replacement cation. Experiments in Ca-free media and in Na-depleted cells indicated that neither phase was due to reversal of Na/Ca exchange. The steep dependence of the Ca<sub>i</sub> increase on Na<sub>o</sub> suggested that the response was not due to lowering intracellular [Na] (Na<sub>i</sub>). The effects of low Na<sub>o</sub> on Ca<sub>i</sub> were also completely independent of changes in intracellular pH (pHi). Cai was remarkably stable during changes of pH<sub>i</sub> of up to 2 pH units, indicating that H and Ca do not share a cytoplasmic buffer system. Such large pH excursions required determination of the pH dependence of fura-2. Because fura-2 was found to decrease its affinity for Ca as pH decreased below 6.7, corrections were applied to experiments in which large pH<sub>i</sub> changes were observed. In contrast to the relative insensitivity of  $Ca_i$  to changes in  $pH_i$ , decreasing extracellular pH (pH<sub>o</sub>) to 6.0 or below was found to stimulate release of intracellular Ca stores. Increased Ca entry was not observed in this case. The ability of decreases in Na, and pH<sub>o</sub> to stimulate release of intracellular Ca stores suggest interactions between Na and H with extracellular receptors.

Key Words gastric glands · fura-2 · calcium · sodium · pH

#### Introduction

Cytoplasmic free [Ca] (Ca<sub>i</sub>) is a critically regulated component of the cytosol. The concentration of free-Ca ions in the cytoplasm represents a pool of Ca which, in the steady state, reaches equilibrium with intracellular Ca-binding sites, some of which serve regulatory functions. A variety of hormones, peptides and neurotransmitters are believed to exert their effects by altering Ca<sub>i</sub>. In the case of the gastric parietal cell, for example, histamine, gastrin and the cholinergic agonist, carbachol, each cause elevation of Ca<sub>i</sub> (Chew & Brown, 1986). The agonist-stimulated Ca increases are biphasic, consisting of an initial release of Ca from intracellular stores and a sustained Ca influx from the extracellular space (Negulescu & Machen, 1988*a*). The elevation of Ca<sub>i</sub> has an important role in regulating HCl secretion from the parietal cell (Negulescu, Reenstra & Machen, 1989).

In addition to the ability of agonists to raise  $Ca_i$ , it is known that the intracellular concentrations of Na (Na<sub>i</sub>) and H (pH<sub>i</sub>) can affect Ca<sub>i</sub> in some cell types. For example, in excitable cells increasing  $Na_i$ (or removing  $Na_o$ ) causes  $Ca_i$  to rise due to the decreased driving force for Ca extrusion via Na/Ca exchange (e.g., Dipolo & Beauge, 1983). The effects of pH on Ca<sub>i</sub> are more complex: decreasing  $pH_i$  has been observed to increase, decrease, or have no effect on  $Ca_i$ , depending on the cell type. In squid axons and Xenopus oocytes there is an inverse relationship between  $pH_i$  and  $Ca_i$  such that decreasing pH<sub>i</sub> increases Ca<sub>i</sub> (Baker & Honerjager, 1978; Rink, Tsien & Warner, 1980), while increasing  $pH_i$  decreases Ca<sub>i</sub> (Mullins & Requena, 1979). These results suggest reciprocity between intracellular buffer systems for Ca and H. In contrast, decreasing  $pH_i$  in *Helix* neurones resulted in a small Ca, decrease (Alvarez-Leefmans, Rink & Tsien, 1981) while increases in  $pH_i$  result in increases of  $Ca_i$  in lymphocytes (Grinstein & Goetz, 1985) and smooth muscle cells, (Siskind et al., 1989). Finally, alterations of  $pH_i$  had no apparent effect on Ca<sub>i</sub> in fibroblasts (Ives & Daniel, 1987). One complication of studying the relationships between Na and Ca or H and Ca in intact cells is that it is difficult to change Na and pH selectively because  $Na_i$  and  $pH_i$ are linked through Na/H exchange. Since decreases in  $pH_i$  could result in increases in Na<sub>i</sub> and vice versa, it is important to control for the effects of such interactions.

The original intention of this study was to evaluate the role of Na in regulation of Ca<sub>i</sub> in parietal cells from intact rabbit gastric glands. This included testing for the presence of Na/Ca exchange.  $Ca_i$  was measured using the fluorescent, Ca-sensitive probe fura-2. Fluorescence measurements from individual cells within glands were made using microfluorimetry. Removal of extracellular Na (Na $_{a}$ ) caused Ca $_{i}$ to increase. Surprisingly, this increase occurred in both Ca-containing and Ca-free bathing solutions. Furthermore, the Ca increase in Ca-containing solutions was found to be biphasic, consisting of both Ca release from hormone-sensitive internal stores and entry of extracellular Ca. Thus, having set out to investigate a Na-dependent mechanism to lower  $Ca_i$  (Na/Ca exchange), a Na-sensitive mechanism for raising Ca was identified. The release appeared to be independent of Na<sub>i</sub>. Because removal of Na<sub>a</sub> acidifies cells through reverse operation of Na/H (Paradiso, Tsien & Machen, 1987) exchange, we were concerned with identifying possible relationships between  $pH_i$  and  $Ca_i$ .  $pH_i$  was monitored with BCEF or 5-(and 6-)biscarboxyfluorescein diacetate. Decreasing  $pH_i$  by up to 2 pH units at normal  $pH_a$ had very little effect on Ca<sub>i</sub>, even when the pH dependence of fura-2 Ca binding was taken into account. However, lowering extracellular pH (pH $_o$ ) to 6.0 or below caused release of intracellular Ca stores. Thus, lowering either extracellular Na or pH caused changes in  $Ca_i$  which resembled the action of agonists. All evidence indicates that the effects of low Na or pH are mediated by action at some external site.

During the course of this study, two papers appeared which pertained to extracellular Na removal and Ca mobilization. Muallem, Beeker and Pandol (1988) showed that replacement of  $Na_{o}$  with tetramethylammonium (TMA) or choline caused elevation of Ca<sub>i</sub> in gastric glands and pancreatic acini by releasing intracellular Ca stores. Because this effect could be blocked with atropine. it was concluded that the organic cations were binding to cholinergic receptors. More recently Smith, Dywer and Smith (1989*a*) reported that removal of Na<sub>o</sub> mobilizes Ca<sub>i</sub> and triggers inositol phosphate turnover in a variety of cultured cells which do not possess cholinergic receptors. This latter group concluded that  $Ca_i$  elevation could be caused by  $Na_o$  removal itself. In the present study we show that, although parietal cells possess cholinergic receptors which can be stimulated by choline, removal of  $Na_{\rho}$  itself can raise  $Ca_i$ through a mechanism distinct from cholinergic receptor activation. (Portions of this work have appeared in abstract form (Negulescu & Machen, 1988b).)

## **Materials and Methods**

#### MATERIALS

All chemicals were reagent grade and, unless otherwise specified, obtained from Sigma.

## **ISOLATION OF GASTRIC GLANDS**

Gastric glands from New Zealand White rabbits were prepared as described previously (Berglindh & Obrink, 1976). After sacrifice, the stomach was perfused with phosphate-buffered saline (100%  $O_2$ , pH 7.4) via retrograde perfusion of the descending aorta. This facilitated separation of the gastric mucosa from the underlying muscularis. After mincing, the mucosa was placed and stirred in a digestion medium that contained 0.3 mg/ml Type 1A Collagenase in an Eagle's minimum essential medium (Gibco, Grand Island, NY), supplemented with bovine serum albumin (BSA) (Calbiochem, San Diego, CA),  $10^{-4}$  M cimetidine (Sigma, St. Louis, MO) and 20 mM N-2-hydroxyethyl-piperazine-N'-2ethanesulfonic acid (HEPES) (pH 7.45, 100%  $O_2$ , 37°C). Glands were formed within 45 min. These isolated glands were allowed to settle and were then washed several times at room temperature in the same medium without enzymes or albumin.

#### Dye Loading and Calibration

For Ca<sub>i</sub> measurements, a suspension of isolated gastric glands (5% cytocrit) was loaded with fura-2/AM (Molecular Probes, Eugene, OR) in Eagle's medium containing  $10^{-4}$  M cimetidine and 10  $\mu$ M dye for 30 min at 24°C. Following loading, the glands were washed in Eagle's medium and left at room temperature until use. Excitation light was provided by two monochomators set at 340 and 385 nm which were rapidly alternated by a rotating chopper mirror and passed to the sample via a 395 dichroic. The emitted light passed through a UV filter (450 nm long-pass) before counting.

Ca<sub>i</sub> was calibrated in each cell using the formula derived by Grynkiewicz, Poenie and Tsien (1985) for dual-wavelength measurements.

$$Ca_i = K(R - R_{min})/(R_{max} - R)$$
(1)

where  $R_{\min}$  is the ratio of fluorescence intensities at 340 and 385 nm obtained at 0 Ca,  $R_{\max}$  is the ratio at saturating Ca, and R is the measured ratio. At the end of each experiment  $R_{\min}$  and  $R_{\max}$  were determined by treating cells with 10–15  $\mu$ M digitonin in the presence and absence of CaCl<sub>2</sub>. Digitonin was used because the  $R_{\max}$  achieved in this manner was higher than that achieved by the Ca ionophore, ionomycin. In Eq. (1) K represents  $K_d(F_{\min}/F_{\max})$ , where  $K_d$  is the dissociation constant for fura-2 under the appropriate conditions and  $F_{\min}$  and  $F_{\max}$  are the fluorescence intensities at 385 nm minus and plus Ca, respectively. The effective dissociation constant for fura-2 in parietal cells has been calculated to be 300 nm at 37°C, pH 7.0 (Negulescu et al., 1989).

Because fura-2 is a tetracarboxylic acid it is possible that its affinity for Ca is sensitive to H. In the original description of the dye only a small change in fluorescence was noted when pH was changed from 7.05 to 6.75 in the presence of 200 nM free Ca. Because  $pH_i$  approached 6.0 in some of our experiments, we



**Fig. 1.** Fura-2 sensitivity to pH in vitro. (A) Ca titrations in the presence of 5  $\mu$ M fura-2 are shown at pH 7.0, 6.5 and 6.0 as described in Materials and Methods. Since the 340/385 nm ratio at saturating Ca decreased as pH decreased, the ratios obtained at different Ca values are expressed as a percentage of the maximum ratio so that the curves are comparable. (B)  $K_d$  for fura-2 (calculated as in Materials and Methods) as a function of pH

extended the characterization of the pH sensitivity of fura-2. Ideally, this characterization would be performed on dye in cells. However, we felt the Ca<sub>i</sub> titration, which required precise control of both Ca<sub>i</sub> and pH<sub>i</sub> could not be performed on parietal cells. Therefore, the pH dependence of fura-2 between pH 7.5 and 6.0 was evaluated by in vitro titration of fura-2 free acid in cuvettes. The experimental conditions were similar to those described by Grynkiewicz et al. (1985). Briefly, a solution containing 100 mM KCl, 5 mM MOPS, 5 mM MES, 10 mM K<sub>2</sub>H<sub>2</sub>EGTA and 2  $\mu$ M fura-2 was titrated to a given pH and then serially diluted with an identical solution in which K<sub>2</sub>H<sub>2</sub>EGTA was replaced with 10 mM CaH<sub>2</sub>EGTA. Excitation spectra were obtained at each dilution using a SPEX fluorolog fluorimeter (emission 505 nm). Because the solutions contained no Mg, calculation of free Ca at each dilution was given by

 $[Ca] = K_d([CaEGTA]/[EGTA])$ 

in which  $K_d$  is the dissociation constant of EGTA at the pH of the titration. The  $K_d$ s for EGTA at various pH's were given by Tsien and Pozzan (1989). An indication of the effect of pH on fura-2 can be seen in Fig. 1*A*, where the ratios of fluorescence intensities at 340/385 are plotted as a function of Ca. It is apparent that decreasing pH<sub>i</sub> shifts the sensitivity of the probe to the right, which indicates decreased affinity for Ca at low pH.

In order to calculate the  $K_d$  for fura-2 at a single pH at least three ratios are necessary—at 0 Ca, saturating Ca, and some intermediate Ca. With these values, the  $K_d$  for fura-2 can be calculated by rearrangement of Eq. (1).

$$K_{d} = [Ca]/(R - R_{min}/R_{max} - R)(F_{min}/F_{max})$$
(3)

where [Ca] is the free [Ca] given by Eq. (2). Calculated in this manner the  $K_{as}$  for fura-2 at pH 7.5, 7.0, 6.5 and 6.0 were 105, 117, 135 and 248 nM, respectively. These values are plotted in Fig. 1*B*.

Although these values were obtained at room temperature in free solution, and therefore are different from absolute values which would be obtained in cells, the basic pH sensitivity of the dye is unlikely to be altered by the intracellular environment. Therefore, an approximation of the effect on the dye in cells can be made. In order to apply a correction for pH to the calibration of Ca<sub>i</sub> the % change of  $K_d$  from pH 7.0 was calculated from Fig. 1b and then multiplied by the apparent  $K_d$  of the dye in cells at pH 7.0 (300 nM). For example, if pH<sub>i</sub> changed from 7.0 to 6.0, one would predict an 83% increase in the  $K_d$  (i.e., from 300 to 549 nM). Ca<sub>i</sub> can now be calculated based on the new  $K_d$ . In experiments where pH<sub>i</sub> changed by more than 0.3 pH units the corrected Ca<sub>i</sub> is shown as a dashed line.

For measurements of pH<sub>i</sub>, glands were incubated with either 2  $\mu$ M BCECF/AM or 15  $\mu$ M 5-(and 6-)biscarboxyfluorescein diacetate (Molecular Probes) for 30 min at room temperature. BCECF (pK<sub>a</sub> 7.0) was used when pH<sub>i</sub> was expected to remain above 6.4 while fluorescein diacetate (pK<sub>a</sub> 6.3) was used to monitor pH<sub>i</sub> less than 6.4. Excitation of the fluorescein-based indicators was alternated between 440 and 490 nm (510 dichroic). Emitted light above 520 nm was collected. Calibration of pH<sub>i</sub> was performed in a nigericin-containing (20  $\mu$ M) high K solution at various pH<sub>a</sub>'s.

#### MICROFLUORIMETRY

 $Ca_i$  or pH<sub>i</sub> was measured at 37°C in single parietal cells within intact rabbit gastric glands using microspectrofluorimetry dyeloaded glands (see above) were settled onto a cover glass, which was mounted into a temperature-controlled rapid perfusion chamber. The chamber was placed on the stage of an inverted microscope (Zeiss IM35), and a 63× oil objective (Zeiss Neofluar 1.25NA) permitted identification of parietal cells in intact gastric glands. Excitation of the dye was achieved using a modular fluorimeter system (Spex fluorolog 2, model F2C) as described previously. Fluorescence emission from a single cell was obtained using an image plane pinhole which corresponded to a spot diameter of 20 µM. Emission intensity from a parietal cell was measured using a photomultiplier attached to a Spex Datamate microcomputer which stored the two alternating fluorescence signals in separate memories. Data points were recorded every second.



**Fig. 2.** Removal of extracellular Na (NMG replacement) causes an increase of Ca<sub>i</sub> in parietal cells. (A) Na removal elevates Ca<sub>i</sub> in both Ca-containing and Ca-free (0.1 mM EGTA) solutions. (B) Relative contributions of intracellular and extracellular Ca to Nafree-induced increase in Ca<sub>i</sub>. Na-free solutions appear to release internal stores and increase Ca entry. (C) Increases in Ca<sub>i</sub> are independent of the replacement cation for Na. Increases were seen in Na-free solutions containing either 150 mM K, 150 mM Cs, or 150 mM Li

### Solutions

Experiments were performed in a Ringer's solution containing (in mM): 150 NaCl, 2.5 K<sub>2</sub>HPO<sub>4</sub>, 1.0 CaCl<sub>2</sub>, 1.0 MgSO<sub>4</sub>, 10.0 glucose, 10.0 HEPES/NaOH, pH 7.45. In Na-free solutions, Na was replaced with either 150 mм N-methyl-D-glucamine (NMG), 150 mм K or 150 mм Li or 150 mM Cs. In low Na solutions NaCl was replaced isosmotically with NMG-Cl. For example, a solution containing 20 mM NaCl consisted of (in mM): 130 NMG-Cl, 20 NaCl, 2.5 K<sub>2</sub>HPO<sub>4</sub>, 1.0 CaCl<sub>2</sub>, 1.0 MgSO<sub>4</sub>, 10.0 glucose, 10 HEPES/NMG, pH 7.45. NMG-Cl stock solutions were made by weighing out enough NMG free base to make 100 ml of 1 M NMG. Then a 1 N HCl solution was used to dissolve the NMG base powder and bring the volume to 95 ml. pH was then titrated to 7.0 using a concentrated HCl solution and brought to 100 ml with H<sub>2</sub>O. For NH<sub>4</sub> pulse experiments, NH<sub>4</sub>Cl replaced NaCl. The high K calibration solution for BCECF calibration consisted of (in mM): 100 K-gluconate, 30 KCl, 10 Na-gluconate, 10 HEPES/NaOH, 1.0 CaCl<sub>2</sub> and 1.0 MgSO<sub>4</sub>.

## Results

EFFECT OF LOW Na<sub>o</sub> ON Ca<sub>i</sub> IN PARIETAL CELLS

Figure 2 shows the effect of removing extracellular Na (Na<sub>o</sub>) on a single parietal cell within an intact gastric gland. Figure 2A compares the responses obtained in Ca-containing (1 mM) and Ca-free (0.1 mM EGTA). Within 30 sec of Na removal Ca<sub>i</sub> began to increase rapidly and reached similar levels in both Ca-containing and Ca-free solutions, indicating release of Ca from intracellular stores. On average, total removal of Na<sub>o</sub> caused Ca<sub>i</sub> to increase from a resting level of  $93 \pm 12$  nM to  $427 \pm 24$  nM. Following the peak, Ca<sub>i</sub> decreased back toward baseline. In Na-free, Ca-containing solutions Ca<sub>i</sub> usually did not return all the way to control levels, suggesting that Ca may be entering the cell from extracellular space.

Figure 2B shows the relative contributions of intracellular and extracellular Ca to the Ca, increase stimulated by Nao removal. First, a cell was exposed to a Ca-free, Na-free solution. This caused a transient increase in Ca<sub>i</sub>. Then, after Ca<sub>i</sub> had decreased to close to resting levels, Ca<sub>a</sub> was introduced in the continued absence of Na<sub>o</sub>. This caused a second, sustained elevation of Ca<sub>i</sub>. Under these conditions elevation of  $Ca_i$  in the presence of  $Ca_a$ cannot be due to the reverse operation of Na/Ca exchange because Nai had been depleted due to prolonged exposure (at least 10 min) to Na-free solutions. Previous measurements of Na, in parietal cells using the fluorescent dye, SBFI, have shown that this length of time is sufficient to deplete cells of  $Na_i$  (Negulescu et al., 1990). Ca<sub>i</sub> returned back to baseline upon readdition of Na<sub>o</sub>. These results indicate that removal of Na<sub>o</sub> causes both release of Ca from internal stores and entry of extracellular Ca.

In order to control for possible effects of Na removal on agonist receptors as described by Muallem et al. (1988), control experiments were conducted in the presence of 10  $\mu$ M atropine and 100  $\mu$ M cimetidine. These antagonists block the effects of cholinergic and histaminergic stimuli on Ca<sub>i</sub> (Chew & Brown, 1986; Negulescu & Machen, 1988a). We found that there was no difference in the Na-free responses in the presence of the antagonists (*not shown*). Although most experiments were conducted in NMG-containing solutions, similar results were obtained by replacing Na<sub>o</sub> with K, Cs, or Li (Fig. 2C).

The nature of the released Ca was investigated by sequentially exposing parietal cells to Na-free and carbachol-containing solutions and vice versa. As shown in Fig. 3, if the cell was exposed to Na-



**Fig. 3.** Na-free solutions release Ca from hormone-sensitive intracellular Ca stores. Pretreatment of parietal cells with Na-free solutions attenuated Ca<sub>i</sub> increases due to carbachol (100  $\mu$ M), while treatment with carbachol prevents increases due to Nafree solutions. These experiments were conducted in Ca-free solutions to selectively observe effects on intracellular Ca stores

free solutions first, the Ca<sub>i</sub> response to carbachol was small. Conversely, if the cell was treated with carbachol first, Na-free had no effect. These experiments were conducted in Ca-free solutions to prevent refilling of the internal store between treatments (Negulescu & Machen, 1988c) and to rule out the contributions of Ca entry mechanisms to the response. These findings suggest that Na-free and carbachol are releasing Ca from the same intracellular store.

The response to Na-free solutions in Figs. 2 and 3 involved the complete removal of  $Na_0$ . To test the sensitivity of the response to various levels of Na<sub>0</sub>, the concentration dependence of the response was studied. Figure 4A shows a typical experiment in which varying amounts of  $Na_{a}$  were removed from the perfusate. Generally, no effects were seen when Na<sub>o</sub> was greater than 20 mm. However, at 15 mm Na<sub>o</sub> or lower a rather sharp concentration dependence was noted. For the cell shown in Fig. 4, 20 mm Na $_{0}$  caused only a very slight increase in Ca, whereas 10 mM Na<sub>o</sub> produced a maximal effect. The Ca<sub>i</sub> response to a maximal dose of 100  $\mu$ M carbachol is shown for comparison. Figure 4B shows the average peak Ca, achieved by various Na<sub>a</sub>. Similar results were obtained in cells that had been treated with carbachol (e.g., Fig. 4A) and in cells that had never been exposed to carbachol (e.g., Fig. 2A).

Initially, removal of  $Na_o$  might be expected to hyperpolarize the cell due to the loss of Na through conductive pathways. A secondary depolarization would occur due to eventual inactivation of the Na/K ATPase. The effects of both depolarization and hyperpolarization of the membrane potential



Fig. 4. Na-dependence of the Ca<sub>i</sub> response to low Na solutions. (A) Ca<sub>i</sub> increases where only detected when Na<sub>o</sub> at 20 mM or below. Response to 100  $\mu$ M carbachol shown for comparison. (B) Na concentration dependence of Ca<sub>i</sub> response to low Na. Values represent averages (±SEM) for cells exposed only once, at a single concentration of Na<sub>o</sub>. Averages based on responses from at least five cells in two separate preparations. Half-maximal responses were obtained at 16 mM Na<sub>o</sub>

were investigated by altering external K. Neither high K (50 mM) nor low K (0.5 mM) had any significant effect on Ca<sub>i</sub> (not shown).

Ca<sub>i</sub> Increases upon Na<sub>o</sub> removal Are Independent of Changes in  $pH_i$ 

Gastric parietal cells are known to possess an Na/ H exchanger which normally functions to extrude H from the cytoplasm and is responsible for pH<sub>i</sub> recovery following acid loads in HEPES-buffered (HCO<sub>3</sub>-free) solutions (Paradiso et al., 1987). Removal of Na<sub>o</sub> results in the reverse operation of the exchanger and causes pH<sub>i</sub> to acidify. Acidification of the cytoplasm elevates Ca<sub>i</sub> in squid axon (Baker & Honerjager, 1978). Therefore, it was important to determine to what extent changes in Ca<sub>i</sub> correlated with changes in pH<sub>i</sub> when Na<sub>o</sub> was removed. This was done by performing parallel experiments in fura-2 and BCECF-loaded cells. In order to distinguish between the effects of pH<sub>i</sub> and Na<sub>o</sub> on Ca<sub>i</sub>, cells were acidified using maneuvers which did not



**Fig. 5.** Ca<sub>i</sub> increases caused by Na removal are independent of changes in  $pH_i$ .  $pH_i$  was acidified by pulsing the cells with either 5% CO<sub>2</sub>, 20 mM NH<sub>4</sub>Cl, or Na removal. Na removal caused the largest increase, which occurred before  $pH_i$  had dropped by more than 0.10 pH units.  $pH_i$  and Ca<sub>i</sub> were measured in parallel experiments

involve removal of Na<sub>o</sub>. Figure 5 shows that acidification of cells, produced by either a pulse of 5% CO<sub>2</sub> or 30 mM NH<sub>4</sub>, had small effects on Ca<sub>i</sub> when compared to the effect of Na<sub>o</sub> removal. Taking the pH-dependent affinity shift of fura-2 into account, Ca<sub>i</sub> increased by 15% at pH 6.5 (i.e., increased by 20 nM). In contrast to these relatively small changes, removal of Na<sub>o</sub> increased Ca<sub>i</sub> to 380 nM. In addition, note that increases in Ca<sub>i</sub> due to Na<sub>o</sub> removal occurred before pH<sub>i</sub> had decreased by more than 0.05 pH units. Thus, while large decreases in pH<sub>i</sub> elicit small increases in Ca<sub>i</sub>, it does not appear that pH<sub>i</sub> contributes to the rapid increases in Ca<sub>i</sub> due to Na-free solutions.

## EFFECT OF LOW $pH_o$ on $Ca_i$

Another way to acidify cells is to lower the pH of the external solution. In contrast to the maneuvers shown in Fig. 5, in which  $pH_i$  was changed at constant  $pH_o$ , decreasing  $pH_o$  caused large increases of  $Ca_i$ . Figure 6A shows that decreasing pH<sub>o</sub> from 7.4 to 5.7 or lower caused progressive increases in  $Ca_i$ . No effect was detected at  $pH_o$  above 6.2. The Ca<sub>i</sub> increase was due entirely to release of intracellular Ca. Note that there was a lag of about 45 sec between the time  $pH_o$  was decreased and the time of the response (Fig. 6B). Complete release of the internal store, assessed by the ability of carbachol to cause additional Ca release, was observed at a  $pH_o$ of about 5.0. Figure 6C shows the dose response for external protons. The Ca<sub>i</sub> increases were due to release from hormone-sensitive internal stores, since the response was observed in Ca-free solutions (Figs. 6A and B and 7) and was blocked by pretreatment of carbachol (Fig. 7). Pretreatment of cells with atropine had no effect on the ability of low pH to increase  $Ca_i$  (*not shown*).

Data presented in Fig. 5 show that lowering  $pH_i$ to 6.7 had only small effects on  $Ca_i$ . However, when  $pH_a$  was lowered to 6.0 or below,  $pH_i$  was expected to drop rapidly below 6.7. To determine  $pH_i$  at the time of the Ca elevation,  $pH_i$  was measured with fluorescein diacetate, which has a  $pK_a$  of 6.3 and is appropriate for measuring more acidic pH<sub>i</sub>'s. Figure 8 shows parallel experiments in which  $pH_i$  and  $Ca_i$ were measured when cells were subjected to a large acidification using either 50 mM NH<sub>4</sub> or by lowering  $pH_o$  to 5.5. A pulse of 50 mM NH<sub>4</sub>Cl followed by incubation in 250  $\mu$ M amiloride and 500  $\mu$ M H<sub>2</sub>DIDS (which blocked the pH regulatory response) acidified cells to  $pH_i$  6.0, which had minimal effects on the fura-2 ratio. Once corrected for decreased affinity of fura-2 at low pH (Fig. 1),  $Ca_i$  still only increased to 200 nm. Thus, the effect of low  $pH_a$  appears to be largely due to an effect of H at an external site.

## INTERACTIONS BETWEEN LOW Na<sub>o</sub> AND pH<sub>o</sub>

Figure 9A shows that  $Ca_i$  increases resulting from 20 mM Na<sub>o</sub> and pH<sub>o</sub> 5.7 are additive. An interesting finding was that while ouabain by itself had no effect on Ca<sub>i</sub>, it augmented Ca<sub>i</sub> increases induced by low pH<sub>o</sub> (Fig. 9B). Since the transmembrane Na gradient appeared to affect the response to pH<sub>o</sub>, one possibility was that these effects were mediated through the Na/H exchanger. However, neither amiloride nor alterations in pH<sub>i</sub> had any effect on the effect of low Na<sub>o</sub> (Fig. 10), which indicates that Na/H exchange is not directly responsible for Ca<sub>i</sub> increases.

### Discussion

Lowering Na<sub>o</sub> to less than 10–20 mM rapidly causes a biphasic increase in Ca<sub>i</sub> by releasing internal Ca stores and stimulating Ca influx in parietal cells. Release of internal stores was shown by Na-freeinduced increases in Ca<sub>i</sub> in the absence of Ca<sub>o</sub> (Fig. 2). Ca entry was observed in Na-depleted cells and therefore was not due to Na/Ca exchange operating in reverse. Because removal of Na<sub>o</sub> increased Ca influx, it was difficult to determine directly whether Na/Ca exchange exists in parietal cells. However, the fact that reducing the transmembrane gradient for Na entry (by lowering Na<sub>o</sub> to 30 mM or by exposing cells to ouabain) had no effect on Ca<sub>i</sub> suggests that Na/Ca exchange does not contribute sig-



nificantly to Ca regulation under these conditions. This agrees with the conclusions of Muallem and Sachs (1985). With regard to Ca<sub>i</sub> metabolism, these effects of low Na<sub>o</sub> are similar to the effects of carbachol or histamine in that Na-free (NMG-containing) solutions were found to release Ca<sub>i</sub> from a carbachol-sensitive pool (Fig. 4) and cause Ca entry (Fig. 2*B*).

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The effects of Na-free solutions on  $Ca_i$  were not due to interactions of the replacement cation with cholinergic or histaminergic receptors since the response persisted in the presence of either atropine or cimetidine (an H2-histaminergic receptor blocker). In addition,  $Ca_i$  increased in solutions where Na was replaced with either K, Cs, or Li, none of which would be expected to stimulate hormone receptors (although Li is known to affect phosphoinositide metabolism). It should be noted, however, that cholinergic receptors can be stimulated if the replacement cation is choline or TMA as shown by Muallem et al. (1988) and confirmed in our own control experiments (*not shown*).

While the data indicate that the increase in  $Ca_i$ upon Na removal is not due to interactions with cholinergic or histaminergic receptors, our observa-



**Fig. 7.** Low  $pH_o$  releases Ca from carbachol-sensitive intracellular stores. Experiments were conducted in Ca-free solutions to selectively observe effects on intracellular stores. Ca<sub>i</sub> increases due to carbachol were markedly attenuated if Ca stores had previously been released by perfusing cells with solution at pH 5.5. Conversely, if parietal cells were first treated with carbachol, Ca<sub>i</sub> responses to low pH were blocked

tions are nevertheless most consistent with action at some external site. The  $Ca_i$  increase after  $Na_o$ removal was rapid, exhibited a steep dose dependence, and was independent from effects on mem-





**Fig. 9.** Interactions between low pH and low Na. (A)  $Ca_i$  increases caused by low Na and pH are additive. (B) Raising  $Na_i$  increased the effectiveness of low pH

ouabain

pH 5.5

pH 5.5



brane potential (as elicited by 10-fold increases or decreases of  $K_o$ ), and pH<sub>i</sub> (elicited by pulsing cells with CO<sub>2</sub> or NH<sub>4</sub>Cl). Since volume or Na<sub>i</sub> changes are probably similar at 30 mM (which had no effect on Ca<sub>i</sub>) and 20 mM Na<sub>o</sub> (which did), it is unlikely that either Na<sub>i</sub> or volume are triggers for Ca<sub>i</sub> increases following Na<sub>o</sub> removal. Although not shown in the results, similar responses to low Na and pH were obtained in chief cells, which also possess hormone-sensitive Ca stores (Chew & Brown, 1986). The similarity between the responses of these two cell types and the similarity of the low Na results with those of Smith et al. (1989*a*) suggest that this may be a feature common to many cell types.

Precedents exist for several possible mechanisms to explain the effects of low Na<sub>o</sub> on Ca metabolism. Smith et al. (1989*a*) have suggested that a Na-sensitive "receptor" is responsible for triggering release of internal Ca stores, based on the observation that Na removal triggers phosphoinositide breakdown. If such receptors exist, they would need to be sensitive to the loss of Na and their physiological relevance with regard to Ca mobilization is unclear since cells would never be exposed the large decreases in Na<sub>o</sub> needed to elicit a measurable Ca<sub>i</sub> rise. With regard to Ca entry across the plasma membrane, Na-free solutions have been ob-



Fig. 10. The Na-free effect is unaffected by amiloride and independent of pH<sub>i</sub>. To determine whether low Na<sub>a</sub> effects were dependent on Na/H exchange or pH<sub>i</sub>, Na-free solutions were exposed to cells which were held acidic following a 30 mM NH<sub>4</sub>Cl pulse with 100  $\mu$ M amiloride. Under these conditions Na-free solutions elicited a typical response, despite different pH<sub>i</sub> conditions

served to increase Ca influx in phorbol ester-stimulated neutrophils (Nasmith & Grinstein, 1987). In this case removal of Na<sub>o</sub> may have opened Na-sensitive Ca channels. It is possible that either or both of these mechanisms are responsible for the effects on Ca<sub>i</sub> in gastric cells. As pointed out by Smith et al. (1989*a*), the apparently direct effects of Na removal on Ca<sub>i</sub> in the present study are distinct from findings that Na can affect the affinity of agonists and antagonists for their receptors (Motulsky & Insel, 1983; Nunnari et al., 1987), since agonists were not required to observe Ca<sub>i</sub> increases.

The effect of low pH<sub>o</sub> on Ca<sub>i</sub> was similar in several ways to the effect of low Na<sub>o</sub>. For example, decreases in pH<sub>o</sub> produced a rapid (although delayed)  $Ca_i$  increase which was due to release from internal stores. Second, the response exhibited a steep "dose response." Also, the effect of H was due mostly to action at some external site since drastic excursions of pH<sub>i</sub> in either alkaline or acidic directions had little apparent effect on Ca, even when the decreased affinity of fura-2 for  $Ca_i$  at pH values below 6.5 were considered. This independence implies the lack of a shared cytoplasmic buffer system between H and Ca. There were two differences between the effects of low  $pH_a$  and low  $Na_o$ . First, the sensitivity to pH occurred as [H] increased, whereas the Na response was observed as Na concentrations decreased. A second difference was that low  $pH_{o}$  did not seem to enhance Ca entry from outside the cell.

The physiological significance of the response to low  $pH_o$  is unclear, since it occurred at pH values lower than the basolateral membranes of these cells are likely to encounter under normal physiological conditions. However, it may reflect a general phenomenon since it has also been observed in fibroblasts (Smith, Dwyer & Smith, 1989b).

The finding that both low pH and low Na stimulated  $Ca_i$  increases prompted the suggestion that the effects could be mediated through the same mechanism. One such candidate would be the Na/H exchanger, especially since it would be expected to reverse directions at about 20 mM Na<sub>o</sub>. This is due to equilibrium conditions of the exchanger described by

$$Na_i/Na_o = [H_i]/H_o] = antilog(pH_i - pH_o).$$
 (4)

Since  $Na_o = 150 \text{ mM}$ ,  $pH_o = 7.4$  and  $pH_i = 7.1$ under control conditions, the exchanger will extrude protons at normal  $Na_i$ , which has been determined to be 10 mM (Negulescu et al., 1988*c*, 1990). However, at 20 mM  $Na_o$  or below, the exchanger should reverse. The finding that ouabain treatment of cells augmented  $Ca_i$  release due to low  $pH_o$  (Fig. 9*B*) is consistent with an effect on the Na/H exchanger since the transmembrane gradient of Na would be smaller, resulting in more reverse flux through the exchanger at low  $pH_o$ . However, the lack of blockade with amiloride (Fig. 10) argues against such a mechanism and in favor of a "receptor."

That cells are sensitive to low Na<sub>o</sub> or pH<sub>o</sub> has implications for experiments in which the Na gradient is altered to assess the activity of Na-dependent transporters or where pH<sub>o</sub> is altered to study pHdependent processes. For example, one must consider the possible effects of stimulating Ca<sub>i</sub> metabolism, including emptying of intracellular Ca stores, when investigating Na-dependent transporters, particularly Na/Ca exchange. Ideally, such experiments would be conducted at sufficiently high Na or pH as to avoid complications due to the phenomena described here.

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